

Systematic Analysis of Quantum Dot Synthesis in Reverse Micelles

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Background

Quantum dots (QDs) are nanoscale semiconductor crystals with optoelectronic properties that have a variety of applications. The optoelectronic properties of quantum dots are closely related to their size. Therefore, the fabrication method of quantum dots is crucial in producing dots with specific properties.



Figure 1: Size-tunable emission spectra of quantum dots [1]

RM Exchange and the Population Balance Model



Based on this model, variables to be tested were chosen: Time Viscosity of organic Precursor Amount • Water loading (W₀)

Trend Analysis

The locations and intensity counts of the seed and quantum dot peaks were extracted from the EEMS data using a MATLAB script, then compiled with W_o viscosity of organic, and time point to run a data mining statistical analysis in R.



from trend analysis

References

Water loading: strong (+) correlation between W_0 and QD emission wavelength; reflects literature

Viscosity: very strong (+) correlation between seed intensity and viscosity as well as (-) correlation between QD emission wavelength and viscosity; agrees with expectation that higher viscosity reduces yield of mature QDs due to less inelastic collisions between RMs

Precursor Amount: strong (+) correlation between QD intensity and W₀ in filled series while constant series shows (-) correlation; there is less precursor in the constant system to form the larger QDs as W₀ increases, resulting in a lower yield. Therefore, QD yield can be controlled through amount of precursor in the system.

Using reverse micelles (RMs) to control quantum dot crystal formation has the potential to offer a cost-effective, reproducible, and most importantly, scalable method for quantum dot fabrication. RMs are spontaneously-organizing nanobubbles of water surrounded by surfactant in a nonpolar solvent. RM size is controlled through water loading (molar ratio of surfactant to water)



Figure 2: Reverse Micelle with cargo ^[2]

Excitation-Emission Spectra

Fluorescence data was gathered in the form of excitationemission spectra. These spectra consist of two signals: seeds (from nucleation) and mature quantum dots. EEMS was taken daily for four days, allowing for the growth of the QDs to be observed.

An example series of EEMS spectra is shown in Figure 5. This set allows Ostwald ripening (seeds dissolve and redeposit on larger crystals to form mature QDs) to be observed. Day 0 shows a high intensity of seeds that decreases through Day 3, where there is a higher intensity of QDs.



Figure 4: Ostwald Ripening process of crystal growth



Figure 5: EEMs spectra for filled heptane sample of water loading 4 for Day 0 (left) through Day 3 (right).

EEMS to Size Distributions

In order to apply the population balance model to analyze the data, the EEMS spectra must be converted to size distributions. The Nosaka model for particle size vs. excitation energy was used:[5]



Figure 7: Reproduced graph of Nosaka model

The intensities for each excitation wavelength (200-500 nm) for the location of the QD peak were normalized and used as the frequencies for the corresponding particle sizes.



 $-\frac{-1}{\left(R\sqrt{V_0m_e^*/m_e}+c_1\right)^2}+a_2+\frac{c_2}{\left(R\sqrt{V_0m_h^*/m_e}+c_2\right)^2}\right)$

εR

 b_1



Figure 8: Example size distribution histogram for filled isooctane sample of water loading 5

Next Steps

- · Convert all EEMs spectra to size distributions
- · Apply population balance model to further analyze data

Acknowledgements

Sigma-Aldrich: "Quantum Dots," http://www.sigmaaldrich.com/materials-science/nanomaterials/quantum-dots.html (accessed on 28 February 2019).
Qadir, A., Faiyazuddin, M., Hussain, M. T., Alshammari, T. M., & Shakeel, F. (2016). Critical steps and energetics involved in a successful development of a stable nanoemulsion" *Journal of Malecular Liquids*, 214, 7-18. doi:10.1016/j.mollis.2015.11.050
Fletcher, P. D. L., Hover, A. M., Rohinson, B. H. & H., K. C. T. N. The Kinetics of Solubilizate Exchange between Water Droplets of a Water-in-oil Microemulsion. 985–1006 (1987).
Stot, H. Asaji, N., & Komasawa, I. (2000). A population balance approach for particle coagulation in reverse micelles, 39(2), 328-334. doi:10.102/ie900523j
Nosaka, Y. (1991). Finite Depth Spherical Well Model for Excited States of Ultrasmall Semiconductor Particles. An Application. J. Phys. Chem, 95(13), 5054-5058.

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